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Answer: Question 21. In terms of Charles' law explain why  $-273\text{ }^{\circ}\text{C}$   
is the lowest possible temperature. Answer: At  $-273\text{ }^{\circ}\text{C}$ , volume of the  
gas becomes equal to zero, i.e., the gas ceases to exist. Question 22.  
Critical temperature for  $\text{CO}_2$  and  $\text{CH}_4$  are  $31.1\text{ }^{\circ}\text{C}$  and  $-81.9\text{ }^{\circ}\text{C}$   
respectively. Which of these has stronger intermolecular forces and  
why?

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is working in the lab, with compounds containing carbon. She is most likely working in the field of The answer is A because organic

chemistry is the study of most carbon-containing compounds. \_A\_\_\_\_

2.

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